

KEY - Chem IH Redox Practice

1) +5 NaClO_3 $+1 + x - 6 = 0$

2) -1 NaCl $+1 + x = 0$

3) -4 NaClO_2 $+1 + x - 4 = 0$

4) +7 NaClO_4 $+1 + x - 8 = 0$

5) +2 $(\text{S}_2\text{O}_3)^{-2}$ $2x - 6 = -2$

6) 0 $(\text{C}_2\text{H}_3\text{O}_2)^{-1}$ $2x + 3 - 4 = -1$